Fig. 1 shows that inclusion of $10 \% \mathrm{w} / \mathrm{w}$ methanol or ethanol in acetone casting solvent reduces upper-lower side-difference in permeation rate but increases overall urea permeation rate. Lower methanol or ethanol content had variable effect on rate in a 3 h experiment (Fig. 2). Scanning electronmicroscopy revealed increased surface porosity when methanol or ethanol were included in the casting solvent for both pre- and postpermeation films. These results have obvious significance in the use of films in permeation studies.

## REFERENCES

Anderson, W., Armstrong, P. A. M. \& Abdel-aziz, S. A. M. (1973). J. Pharm. Pharmac., 25, 137P.
Johnston, H. K., \& Sourirajan, S. (1973). J. Appl. polym. Sci., 17, 2485-2499,.
Watt, G. W. \& Chrisp, J. D. (1954). Analyt. Chem., 26, 452-453.

## The relationship between the physical and drug release properties of polyethyleneglycol bases

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Baichwal \& Lohit (1970) have demonstrated that the consistency of theobroma suppository bases can be correlated with drug release. Chiou \& Riegelman (1971) have shown that lipophilic drugs dissolved more rapidly when incorporated in polyethyleneglycols (PEG) and the release rates and blood levels of indomethacin have been shown to be superior from PEG than from conventional lipophilic bases (Kerckhoffs \& Huizinga, 1967). The present work examines the relation between in vitro release of prednisolone and the elastic moduli and breaking strength of solid PEG bases.

Cylindrical blocks of PEG 15 mm in diameter and length were prepared by pouring the appropriate molten base, which was just above its congealing point (Collins, Hohmann \& Zopf, 1957), into a brass mould. The prednisolone ( $0.625 \% \mathrm{w} / \mathrm{w}$ ) was dissolved in the molten base to produced blocks for release testing and water, if required, was added immediately prior to pouring.
Prednisolone release was measured on four similar blocks simultaneously using a rotating basket dissolution apparatus at $37^{\circ}$. 2 litres of distilled water was used as the dissolution medium and 5 ml samples were withdrawn every 120 s and analysed for prednisolone content. The times for $25,50,75$ and $100 \%$ release of drug were determined from the concentration/time curves. The elastic modulus and breaking strength of the blocks were determined at $22^{\circ}$, face on, in a compression unit against a load cell, using a constant strain rate of $60 \mathrm{~mm} \mathrm{~h}^{-1}$.
The samples tested were commercially available PEG 1000, 1540, 4000 and 6000 and intermediate molecular weights were produced by blending PEG 1000 with 4000 and 6000 . A significant linear relation was observed between molecular weight and the times for 25 , 50,75 and $100 \%$ drug release. When water was incorporated into PEG 4000 a linear decrease in release times was observed for increase in water content. Because of the difficulty in casting, blocks containing more than $15 \%$ water could not be tested. The release times also exhibited a linear correlation with the breaking strength of the bases.

The relation between the elastic modulus and the release times was of a complex nature and no apparent trend was obvious. Similar results were observed when the elastic modulus and breaking strength were compared with the PEG molecular weight. It is possible that the lack of correlation in these results may be due in part to the anomalous behaviour exhibited by the elastic modulus of PEG 6000. However, definite trends were apparent when the elastic moduli and breaking strengths of $4000 / 1000$ and $1000 / 400$ mixtures were measured. The former blend produced a linear relation, the latter curvilinear.

## REFERENCES

Baichwal, M. R. \& Lohit, T. V. (1970). J. Pharm. Pharmac., 22, 427-432. Chiou, W. L. \& Riegelman, S. (1971). J. pharm. Sci., 60, 1569-1571.
Kerckhoffs, H. P. M. \& Huizinga, T. (1967). Pharm. Weekblad, 102, 1183-1200.
Collins, A. P., Hohmann, J. R. \& Zopf, L. C. (1957). Am. prof. Pharm., 23, 231-234.

